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### (54) METHOD AND PLANT FOR THE REDUCTION OF THE CONCENTRATION OF POLLUTANTS AND/OR VALUABLE ELEMENTS IN THE WATER

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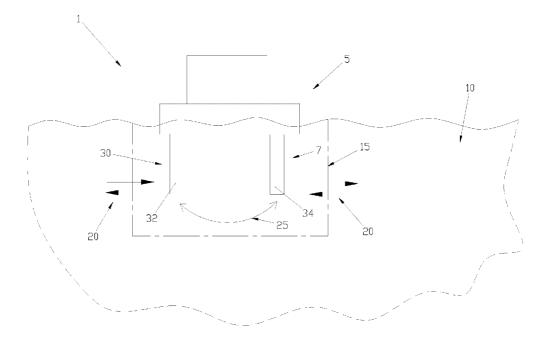
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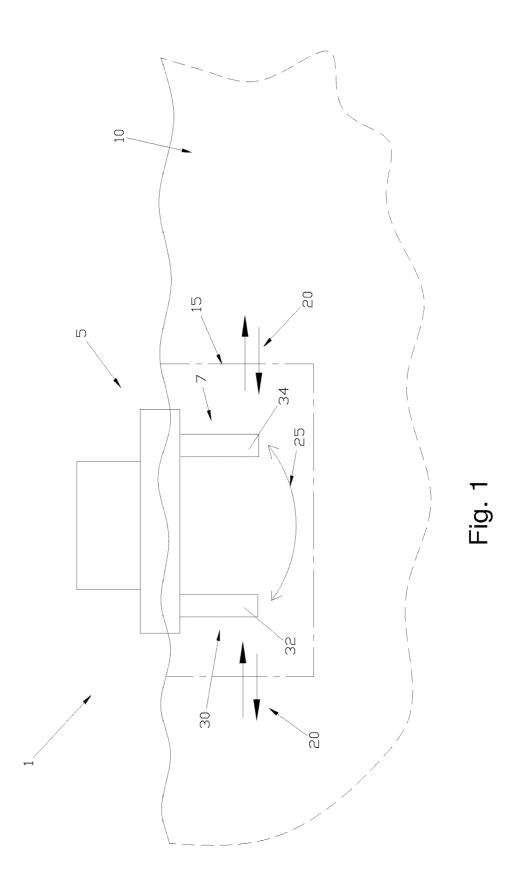
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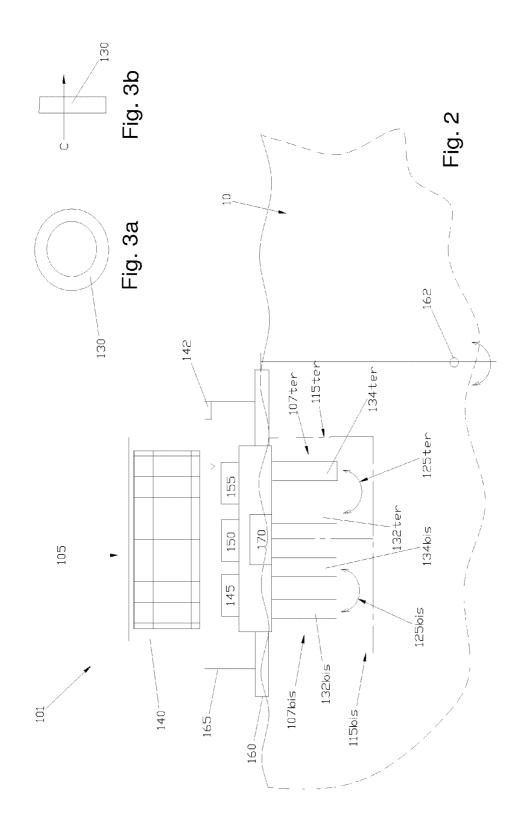
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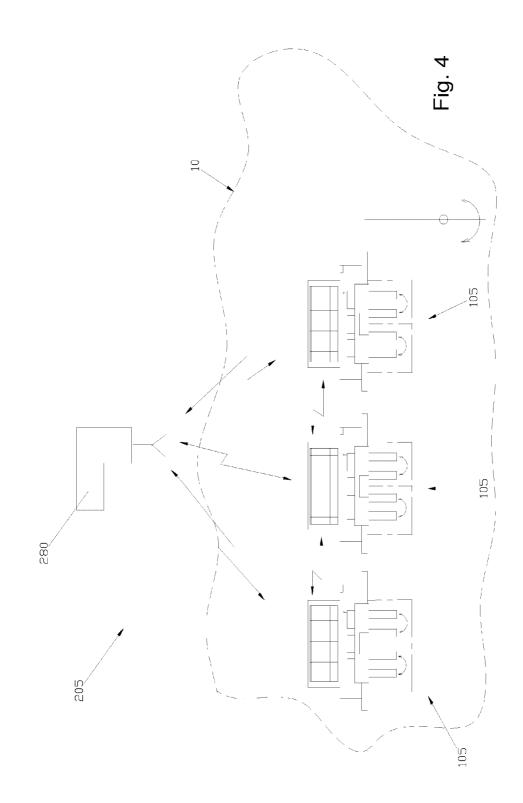
#### (57) ABSTRACT

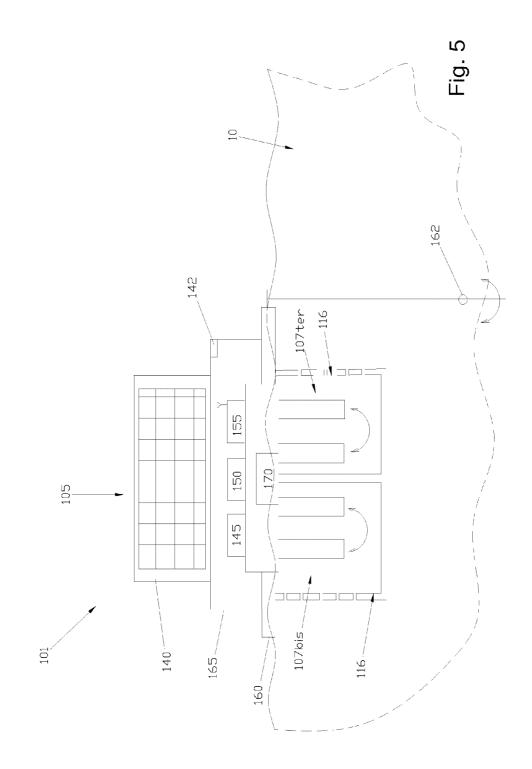
The present invention concerns a method for reducing the amount of polluting and/or valuable elements through application of electrolysis, in particular of the electrocapturing phenomenon. The electrolysis according to the present invention is applied permanently over time in a polluted water body. The predetermined action area (115bis, 115ter) preferably has a smaller extension than the water body. The at least one phenomenon (125bis, 125ter) is preferably powered electrically through production of electrical energy in loco through at least one renewable energy source (140). The method is suitable for purifying large expanses of water, like seas, lakes, lagoons and rivers, through plants operating permanently, however this does not rule out other applications.











#### METHOD AND PLANT FOR THE REDUCTION OF THE CONCENTRATION OF POLLUTANTS AND/OR VALUABLE ELEMENTS IN THE WATER

**[0001]** The present invention concerns a method and a plant for reducing the concentration of polluting and/or valuable elements in water. The present invention has been developed with particular reference to the treatment of large quantities of water so that it is advantageous for use in waters arranged in open system with neighbouring waters, i.e. capable of exchanging at least material with them. However, this does not rule out other applications.

**[0002]** Examples of water treated according to the present invention are areas of free water located in natural basins, like seas, rivers, lakes, lagoons or in artificial basins like dams or canals, or organised water areas capable of flowing from and towards neighbouring areas. These include the water of purification basins of aqueducts and those used in acquaculture or fish farming sites or plants.

**[0003]** As known, the water on our planet is a precious resource for many reasons, which have recently been added to by the fact that unpolluted water is now difficult to obtain.

**[0004]** As known, urban developments, industrial activities and farming activities have a great influence on the pollution of water, in which it is now commonplace to find polluting materials that are harmful to human health suspended in ionic form.

**[0005]** Amongst the most polluting materials we note, in a non-exhaustive list, Cadmium, Lead, Mercury, Aluminium, Antimony, Arsenic, Tin, Thallium and Uranium, many of which are in metallic form.

[0006] These elements, as well as entering our body directly when water is drunk, are taken on through the food chain. A classic example is the concentration mechanism in fish or in acquaculture products that we eat on a daily basis. [0007] Amongst the numerous bibliographic references that we could quote to quantify the extent of pollution of water on the planet and the impact on man we note the following very complete report: ENVIRONMENTAL CON-TAMINANTS ENCYCLOPEDIA, Jul. 1, 1997; COMPIL-ERS/EDITORS: ROY J. IRWIN, NATIONAL PARK SER-VICE; WITH ASSISTANCE FROM COLORADO STATE UNIVERSITY; STUDENT ASSISTANT CONTAMI-NANTS SPECIALISTS: MARK VAN MOUWERIK, LYNETTE STEVENS, MARION DUBLER SEESE, WENDY BASHAM, NATIONAL PARK SERVICE WATER RESOURCES DIVISIONS, WATER OPERATIONS BRANCH; 1201 Oakridge Drive, Suite 250, FORT COL-LINS, COLORADO 80525.

**[0008]** As can be seen, the aforementioned report is from 1997, and yet other previous ones could be given, to demonstrate the fact that the problem has been around for a very long time.

**[0009]** Moving on to the aspect of oncological research, a recent study of the university of Ferrara offered a complete summary of the damage caused in the body by these pollutants: 3.7.2013 Prof. Francesco Caritei, University of Ferrara, Inquinamento della fauna ittica e delle acque da metalli pesanti e conseguenze sanitarie nell' uomo; in particolare in campo oncologico.

**[0010]** The fact that such a report was only a few days prior to the filing date of the present patent application demonstrates the absence of a radical solution to the problem.

**[0011]** It has thus been demonstrated that in current society for many years, there has been a need to find global systems that are applicable on a large scale to reduce the pollution of our planet's water.

**[0012]** Such a requirement is especially great for seas, rivers, lakes and lagoons, and for all water bodies close to sources of pollution.

**[0013]** Also, considering the fact that the mineral resources of the planet are limited, and that such pollutants, as well as other elements present in water, can constitute an important supply source, the usefulness of a purification process could be even greater.

**[0014]** A general purpose of the present invention is therefore to satisfy, at least partially, such a requirement.

**[0015]** A preferred purpose of the present invention is to provide a process for reducing pollutants and/or valuable elements that is easy and cost-effective to carry out and that is applicable effectively on a large scale.

**[0016]** A further preferred purpose of the present invention is to allow recycling of the polluting and/or valuable elements making them re-useable in industry.

**[0017]** According to a first general aspect thereof, the present invention concerns a method for reducing the amounts of polluting and/or valuable elements through application of electrolysis, in particular of the electrocapturing phenomenon. The electrolysis according to the present invention is applied permanently over time in a polluted water body. The invention is based on the ingenious intuition that even by removing small amounts of polluting and/or valuable elements per unit time, it is possible to remove large quantities in the long term from the site to be purified. The method also does not need big interventions performed by man, since it is effective thanks to the long application period.

**[0018]** The method can also advantageously be applied to any polluting and/or valuable element that can have an ionic form, like for example metals and salts. It should be observed that metallic ions and salts in solution can already be present naturally in water and/or can be obtained by electrolytic dissociation, in which case is considered part of the electrocapturing phenomenon.

**[0019]** The following non-exhaustive table shows some polluting and/or valuable elements that can be extracted from water with the method of the present invention:

Element	Total mass in	Production	Mass of water
	the oceans	in 2007	to be processed
	(tonnes)	(Tonnes)	(tonnes)
Li	2.31E+011	2.50E+004	1.40E+011
Mo	1.3E+010	1.87E+005	1.87E+013
U	4.29E+009	6.65E+004	2.02E+013
V	2.47E+009	5.86E+004	3.08E+013
Cd	1.43E+008	1.99E+004	1.81E+014
Au	1.43E+007	2.50E+003	2.27E+014
Sn	1.42±400	3.8E+005	1.07E+015
Ni	3.64±4008	1.76E+006	2.81E+015
Cu	8.23±4008	1.56E+007	1.73E+016
Mn	1.17±4009	1.16E+007	2.90E+016
Zn	5.20±4009	1.80E+008	3.60E+016
Al	1.30±4009	3.80E+007	3.80E+016
Cr	2.60±4008	2.00E+007	1.00E+017
Pb	3.90±4007	2.55E+006	1.18E+017
Fe	4.42±4009	2.25E+009	6.65E+017
Ti	1.17±4006	6.10E+006	6.78E+018
Co	8.84E+006	3.23E+007	9.13E+018

the table in the various columns gives, respectively, an estimation of the total mass of each element contained in the oceans, the amount of metal in question produced through conventional mining techniques and the amount of water that it would be necessary to process in order to be able to extract what is described in the previous column.

**[0020]** The capture can for example be selective, or the selection can even vary over time.

**[0021]** For the purposes of the present invention we consider "an" electrocapturing phenomenon to be that generated by "an" electrolytic cell. For this reason, in the case of use of a plurality of electrolytic cells we will talk of a plurality of such phenomena. Such a plurality of phenomena can be exploited to increase the amount of elements captured per hour and/or to carry out selective capturing, i.e. such that each phenomenon is associated with the capturing of just a few types of predetermined elements, or of even just one type of element, in which case we will talk of differentiated capturing. The element can, for example be selected from those in the table shown above.

**[0022]** The electrocapturing phenomenon comprises at least one electrodeposition phenomenon on at least one electrode (preferably insoluble, for example made from polymeric material in which conductive fibres are immersed, like carbon fibres), however we do not rule out that it may comprise other phenomena, like for example electrolytic dissociation.

**[0023]** According to some preferred embodiments the electrocapturing phenomenon has a power less than or equal to 100 Watt, preferably less than or equal to 50 KWatt, even more preferably less than or equal to 10 KWatt. This allows the phenomenon to be substantially permanently present in the site to be purified without being invasive. Let us consider for example a fish farming or acquaculture site, or a lagoon where there is marine fauna and flora. If the electric currents were too strong they would become actual shocks incompatible with life.

**[0024]** Such a phenomenon can also be supplied with a production of electrical energy in loco through renewable energy sources. This promotes the use of a plant permanently in operation to treat large quantities of water.

**[0025]** The electrocapturing phenomenon with the power indicated is that generated by a single electrolytic cell. In the case in which there are many cells, each of them is meant to generate an electrolysis phenomenon with a power in the indicated limits.

**[0026]** For the purposes of the present invention "electrolytic action area" is used to define that area of the polluted environment to be purified defined by the presence of an electrocapturing phenomenon. Said action area is preferably smaller than the water body.

**[0027]** According to some preferred embodiments each action area substantially corresponds to a single substantially permanent electrocapturing phenomenon, i.e. the action areas do not substantially overlap. According to a preferred convention it is possible to consider the borders of an action area to be those in which the detectable current of an electrocapturing phenomenon becomes less than  $\frac{1}{10}$  of the current detected in the shortest path between the electrodes.

**[0028]** According to some preferred embodiments of the invention the density of the electrocapturing phenomena applied to the polluted water body is comprised between 1 and 10 per square metre.

**[0029]** In general, it should be observed that it is preferable for the action areas to be arranged to each form an open system with at least one neighbouring area of the water body, like for example an adjacent action area or a neutral area (i.e. without electrocapturing phenomenon).

**[0030]** Such an open system is such that the water passes (for example flows) from the at least one neighbouring area to the action area and/or vice-versa in a natural and/or induced manner.

**[0031]** According to a further preferred general characteristic, the electrocapturing phenomena generated each capture a single type of polluting element, so as to promote their collection and their reuse in industry.

**[0032]** The type of element captured can preferably be modified based on the values of at least one characterising parameter involved in the electrocapturing phenomenon, including for example: concentration of at least two polluting and/or valuable elements in the action area, current detected at the electrodes of the at least one electrocapturing phenomenon, available supply current;

**[0033]** According to some preferred embodiments it is possible to rotate the electrodes to arrange them according to orientations particularly favourable for electrocapturing based on the currents with which the water flows in the at least one action area.

**[0034]** According to other preferred embodiments it is possible to vary the intensity and size of the supply currents of the at least one electrocapturing phenomenon over time to influence the ionic mobility of at least one polluting element to be captured.

**[0035]** According to a second general aspect thereof, the present invention concerns a plant for reducing the concentration of polluting and/or valuable elements in water through the application of a method of the type indicated above. It comprises at least one electrolytic cell comprising at least one group of insoluble electrodes able to be immersed in a predetermined area of a polluted water body to generate at least one electrocapturing phenomenon, at least one generator of electric current that exploits at least one renewable energy arranged to power the electrolytic cell, at least one control station programmed to modify the operating parameters of the electrolytic cell that floats, or using a support that can be fixed to the bottom of the water body, like for example a wharf or a stilt house frame.

**[0036]** In general, the preferred embodiments are those in which the cell forms a single self-powered modular unit with the at least one generator and the at least one control station. The electrocapturing device is in the form of a modular unit able to be used by itself or in association with other units. Advantageously, this makes it possible to have substantially self-powered plants with production of electrical energy in loco, the running of which foresees substantially just the maintenance costs, since the cost of the energy supply is substantially zero. Moreover, they can be left to operate for long operating times required to obtain significant capturing results.

**[0037]** Preferably the modular unit has a power less than or equal to 200 KWatt, preferably less than or equal to 100 KWatt, even more preferably less than or equal to 20 KWatt.

**[0038]** According to some preferred embodiments of the invention at least one electrode of the cell is three-dimensional and has a section of greater size with respect to the

other sections, intended to be arranged perpendicular to the direction of the flow of water through the cell to promote deposition.

**[0039]** In general, it should be observed that plants without water holding parts around the electrodes or those that comprise one or more channels to define a predetermined flow of water from and towards the electrodes are preferable.

**[0040]** Some preferred embodiments of the invention comprise a feedback system between the station and at least one device for detecting at least one parameter of the water and/or at least one operating parameter of the at least one cell.

**[0041]** Further characteristics and advantages of the present invention will become clearer from the following detailed description of preferred embodiments thereof, made with reference to the attached drawings and given for indicating and not limiting purposes. In such drawings:

**[0042]** FIG. **1** is a schematic representation of a plant for reducing the concentration of polluting and/or valuable elements according to the present invention with a basic electrolytic device.

**[0043]** FIG. **2** is a schematic representation of a second plant comprising an alternative electrolytic device to that of FIG. **1**.

**[0044]** FIGS. 3*a* and 3*b* are respective front and side schematic views of an electrode of the device of FIG. 2;

[0045] FIG. 4 is a schematic representation of a third plant according to the present invention, e

**[0046]** FIG. **5** is a schematic representation of a variant of the plant of FIG. **2**.

**[0047]** With reference to FIG. **1**, a basic plant for reducing the concentration of polluting and/or valuable elements in water is shown, wholly indicated with reference numeral **1**, which helps to understand the idea forming the basis of the present invention.

**[0048]** The plant **1** comprises a modular device for generating an electrocapturing phenomenon **5** (also simply known as "electrolytic device" or "modular unit") arranged in a water body **10**, for example a marine environment, more specifically a polluted site.

**[0049]** The electrolytic device **5** comprises an electrolytic cell **7** (also simply called "cell"). The cell comprises a group of electrodes **30** powered by electric current and cooperating with one another to generate an electrocapturing phenomenon by electrodeposition on at least one of them of at least one type of polluting element present in the water. The group of electrodes **30** comprises at least two electrodes, i.e. at least one cathode **32** and at least one anode **34**. For the purposes of the present invention the term electrolytic cell is used to define a cell with a single group of electrodes cooperating with each other, i.e. capable of producing a single electrocapturing phenomenon through electrodeposition on one or some electrodes.

[0050] The area of water in which the cell 7 is immersed and in which its electrocapturing phenomenon 25 is present is called "electrolytic action area" 15, and it is smaller than the water body to be purified 10.

**[0051]** The electrolytic action area is an open system, i.e. it exchanges at least water with the rest of the water body **10**, for example by virtue of natural water currents **20**, like in seas, in lakes, in rivers or in lagoons, or by means of forced convection.

**[0052]** Although the electrolytic device 5 shown comprises just one cell 7, nevertheless this does not rule out the possibility of there being a greater number of cells as will be shown

hereafter. The cell **7** shown also comprises just two electrodes, nevertheless this does not rule out the possibility of the cells containing a greater number of them, for example more than one cathode and/or more than one anode electrically connected together in the electrocapturing phenomenon, for example in a bunch.

**[0053]** FIG. **2** represents a second plant **101** comprising a second modular electrolytic device **105**, which differs from that of FIG. **1** in that it comprises two electrolytic cells **107***bis* and **107***ter*. They each generate their own electrodeposition phenomenon, respectively indicated with **125***bis* and **125***ter*. Each electrodeposition phenomenon defines its own action area **115***bis* and **115***ter*. For this reason the electrolytic device **105** is called multiple cell or multiple electrocapturing phenomenon.

[0054] The cell 107*bis* comprises a cathode 132*bis* and an anode 134*bis*, whereas the cell 107*ter* comprises a cathode 132*ter* and an anode 134*ter*.

**[0055]** The electrolytic device **105** also comprises a generator of electrical energy **140** that exploits at least one renewable energy source. The generator **140** represented is a solar panel, but other types of generators are possible, like for example wind generators, those based on wave motion or on tidal power, or combinations thereof.

**[0056]** Preferably, the electrolytic device is also equipped with an auxiliary power socket **142** to allow powering by the mains, for example in the case of an emergency or maintenance of the panel **140**, or for sudden needs.

[0057] The electrolytic device 105 also comprises an accumulator device 145, a control station 150, a telecommunication device 155, a floatation device 160, an insulating screen 165 and a device 170 for detecting at least one parameter of the water and/or at least one operating parameter of the cells. [0058] The accumulator device 145 advantageously accumulates the energy produced by the generator 140 to make it available to the cells 107*bis* and 107*ter* in a more even manner with respect to the production levels in the various hours of the day. However, this does not rule out it being absent and/or a direct power supply from a generator 140 or from a socket 142.

**[0059]** The station **150** is programmed and arranged to modify and manage the operating parameters of the cells based on the data detected by the detection device **170**.

**[0060]** The telecommunication device **155** exchanges data for the station with other electrolytic devices **105** of the same plant, and/or with a remote monitoring station, for example located on solid ground. In this way the control station can modify the operating settings of the electrolytic device **105** also taking into consideration the data relating to other neighbouring electrolytic devices **105** and/or data entered wanted by an external operator. It is also possible to set the operation of the electrolytic device **105** totally manually from the remote station by sending the desired data.

**[0061]** The floatation device **160** is suitable for keeping the electrolytic device **105** afloat, for example it comprises a series of elements made from floating material or a floating platform. The floatation device also comprises a device for anchoring to the bottom of the water body **162**.

**[0062]** The insulating screen **165** is used to keep all of the components of the electrolytic device isolated from the water, thus not being able to get wet.

[0063] The detection device 170 is connected to the control station to allow control in feedback of the electrolytic device 105. The feedback is preferably of the adaptive type.

**[0064]** For example, the detection device **170** detects the intensity and direction of marine currents and the control station commands the rotation of the electrodes, or of at least one of them, to be arranged in the most advantageous manner for them.

[0065] FIGS. 3a and 3b schematically show an electrode 130 with a section that is greater than the others. In particular, the electrode has an annular shape, so that the front section of FIG. 3a has a greater area than the lateral sections of FIG. 3b. Such a greater section can for example be intended to be kept perpendicular to the direction of the marine currents, indicated in FIG. 3b with the arrow C, so as to be hit directly by the flow of water.

**[0066]** FIG. **4** schematically represents a third electrocapturing plant **201** that differs from the plant **101** of FIG. **2** in that it comprises a plurality of electrolytic devices **105**. They preferably are arranged in a network with one another so as to exchange data, and with a remote monitoring station **280** arranged on solid earth through the telecommunication devices **155**.

**[0067]** In use, the plants described here can be used according to a method that foresees the following steps:

- [0068] identifying a water body that is polluted and/or rich in valuable elements 10;
- [0069] arranging, in such a water body 10, at least one electrolytic device 5, 105 with at least one electrolytic cell defining a predetermined electrolytic action area 15, 115*bis*, 115*ter* that is smaller than the water body 10, where such an action area is arranged to form an open system with the neighbouring areas 10, 15, 115*bis*, 115*ter* of the water body;
- **[0070]** supplying the electrolytic cell with at least one electric current to capture at least one type of polluting element, preferably a single type, present in the water in the action area by electrolysis.

**[0071]** In particular, it is possible to carry out the electrocapturing by deposition on at least one electrode of elements suspended in water in the form of ions, including for example Cadmium, Lead, Mercury, Aluminium, Antimony, Arsenic, Tin, Thallium and Uranium, or more generally those shown in the table inserted in the preamble.

**[0072]** The water flows through the plant, and in particular through the electrolytic device, in a natural and/or induced manner, preferably natural.

**[0073]** Preferably, a plurality of electrolytic cells 7, 107*bis*, 107*ter* are provided and they are fed with electric current parameters such that each cell captures a single type of polluting element, the same as or different from the one captured by the other cells. This promotes the collection and reuse of the captured elements.

**[0074]** Preferably, at least one of the following parameters is detected: concentration of at least two polluting and/or valuable elements in the action area, current detected at the electrodes of the electrolytic cell, current available to the generator;

- **[0075]** at least one of the polluting and/or valuable elements detected is selected and the electrolytic cell is set to capture it basing the selection on the at least one parameter detected.
- [0076] the settings of the electrolytic cell are modified to change the at least one polluting element captured each time the at least one detected parameter on which the base the selection exceeds a predetermined threshold value.

**[0077]** In the case in which there is a plurality of cells, it is possible to modify the power supply settings of just one or some or all of the cells to change the type of polluting element captured each time the at least one detected parameter on which to base the selection exceeds a predetermined threshold value.

**[0078]** Thanks to the control station, it is also possible to vary the intensity and size of the power supply currents of the cell (or of the cells) over time to influence the ionic mobility of at least one polluting element to be captured.

**[0079]** As can be noted, up to now we have described embodiments that carry out the selective capture of the polluting and/or valuable elements based on the settings of the electrical parameters, additionally or alternatively the Applicants note that enclosing predetermined areas around each cell with selective porous membranes, i.e. permeable to just one or to predetermined polluting and/or valuable elements is also very effective. An example is illustrated in the variant of FIG. **5**, where the membranes are indicated with reference numeral **116**.

**[0080]** It should also be observed that although the electrocapturing phenomena are carried out on the water in the state in which it is in, it is also possible to condition the water to be treated to increase the efficiency of the process, for example by adding in the action areas at least one element that increases the conductivity of the liquid, like for example sodium chloride.

**[0081]** It should also be observed that preferably the parts that the plants are made up of, where possible, are made from recycled material, so as to increase the eco-compatibility of the invention, and/or from materials inert to electrolysis, so as not to alter the phenomena generated.

**[0082]** Of course, the embodiments and the variants described and illustrated up to now are purely examples and a man skilled in the art, in order to satisfy specific and contingent requirements, can bring numerous modifications and variants, including for example the combination of said embodiments and variants, all in any case covered by the scope of protection of the present invention as defined by the following claims.

1. Method for reducing the concentration of polluting and/ or valuable elements in water characterised in that at least one type of polluting and/or valuable element is captured generating at least one electrocapturing phenomenon by electrolysis (25, 125*bis*, 125*ter*) substantially permanently over time in a water body (10) that is polluted and/or rich in predetermined valuable elements.

2. Method according to claim 1, characterised in that said electrocapturing phenomenon (25, 125*bis*, 125*ter*) comprises at least one electrodeposition phenomenon on at least one electrode (30, 130).

3. Method according to any one of the previous claims, characterised in that a plurality of distinct electrocapturing phenomena (125*bis*, 125*ter*) are applied substantially permanently over time.

4. Method according to any one of the previous claims, characterised in that the electrocapturing (25, 125*bis*, 125*ter*) is carried out selectively.

5. Method according to claim 4, characterised in that the selection varies over time, for example as a function of at least one detected characterising parameter.

6. Method according to any one of the previous claims, characterised in that the at least one electrocapturing phenomenon (25, 125*bis*, 125*ter*), and/or in the case of the pres-

ence of a plurality of electrocapturing phenomena each of them, has a power less than or equal to 100 KWatt, preferably less than or equal to 50 KWatt, even more preferably less than or equal to 10 KWatt.

7. Method according to the previous claim, characterised in that the at least one phenomenon (25, 125*bis*, 125*ter*) is powered electrically through production of electrical energy in loco through at least one renewable energy source (140).

8. Method according to any one of the previous claims, characterised in that said at least one electrocapturing phenomenon (25, 125*bis*, 125*ter*), and/or in the case of the presence of a plurality of electrocapturing phenomena, each of them, is present substantially permanently in, and defines, a predetermined electrolytic action area (15, 115*bis*, 115*ter*).

9. Method according to the previous claim, characterised in that said predetermined action area (15, 115*bis*, 115*ter*) is smaller than the water body (10).

10. Method according to claim 8 or 9, characterised in that each electrocapturing phenomenon (125*bis*, 125*ter*) is present in, and defines, a respective action area (15, 115*bis*, 115*ter*) of the water body substantially different from the area (15, 115*bis*, 115*ter*) in which at least one other electrocapturing phenomenon (125*bis*, 125*ter*) is present.

11. Method according to any one of the previous claims, characterised in that the borders of said action area (15, 115bis, 115ter) are substantially those in which the detectable current of an electrocapturing phenomenon becomes less than  $\frac{1}{10}$  of the current detected in the shortest path between the electrodes that generate it.

12. Method according to any one of the previous claims, characterised in that the density of the electrocapturing phenomena (25, 125*bis*, 125*ter*) applied to the water body is comprised between 1 and 10 per square metre.

13. Method according to any one of claims 8 to 12, characterised in that said action area (15, 115*bis*, 115*ter*) is arranged to form an open system with at least one neighbouring area of the water body (10, 115*bis*, 115*ter*).

14. Method according to claim 13, characterised in that said electrolytic action area (15, 115*bis*, 115*ter*) arranged to form an open system in the water body is arranged so that the water passes from the at least one neighbouring area to the action area and/or vice-versa in a natural and/or induced manner (20).

15. Method according to any one of the previous claims, characterised in that it comprises the step of substantially permanently applying at least two electrocapturing phenomena (25, 125*bis*, 125*ter*) and of supplying each of them with electric current parameters such that each of the two captures at least one type of polluting and/or valuable element different from that or those captured by the other.

**16**. Method according to any one of the previous claims, characterised in that it comprises the step of detecting at least one of the following parameters:

- concentration of at least two polluting and/or valuable elements in the action area, current detected at the electrodes of the at least one electrocapturing phenomenon, supply current available;
- selecting at least one of the polluting and/or valuable elements detected and setting the parameters of the at least one electrocapturing phenomenon (25, 125*bis*, 125*ter*) for its capture basing the selection on the at least one parameter detected,
- modifying the setting of the electrocapturing phenomenon (25, 125*bis*, 125*ter*) to change the at least one polluting

and/or valuable element captured when the at least one detected parameter on which to base the selection exceeds a predetermined threshold value.

17. Method according to any one of the previous claims, characterised in that it comprises the step of detecting at least one characterising parameter of the flow with which the water flows in the at least one action area, and modifying, based on said parameter, the orientation and/or arrangement of at least one electrode (30, 130) present in the action area (15, 115*bis*, 115*ter*) and suitable for generating the at least one electrocapturing phenomenon (25, 125*bis*, 125*ter*).

18. Method according to any one of the previous claims, characterised in that the intensity and size of the supply currents of the at least one electrocapturing phenomenon (25, 125*bis*, 125*ter*) are varied over time to influence the ion mobility of at least one polluting and/or valuable element to be captured.

**19**. Method according to any one of the previous claims, characterised in that the at least one electrocapturing phenomenon (**25**, **125***bis*, **125***ter*) is monitored remotely through a telecommunication system (**155**).

**20**. Method according to any one of the previous claims, characterised in that the at least one polluting and/or valuable element captured by electrolysis is collected and sent to industry for reuse.

21. Plant for reducing the concentration of polluting and/or valuable elements in water through the application of a method according to any one of the previous claims, characterised in that it comprises at least one electrolytic cell (7, 107bis, 107ter) comprising at least one group of insoluble electrodes (30, 130) able to be immersed in a predetermined area of a water body (10) that is polluted and/or rich in valuable elements to generate at least one electrocapturing phenomenon (25, 125bis, 125ter), at least one generator of electric current (140) that exploits at least one renewable energy arranged to supply the electrolytic cell (7, 107bis, 107ter), at least one control station (150) programmed to modify the operating parameters of the electrolytic cell, at least one support device (160) of the electrolytic cell that floats or using a support that can be fixed to the bottom of the water body (162).

22. Plant according to the previous claim, characterised in that it comprises a self-powered modular unit generating at least one electrocapturing phenomenon (5, 105) comprising the at least one electrolytic cell (7, 107*bis*, 107*ter*), the at least one generator (140) and the at least one control station (150).

23. Plant according to the previous claim, characterised in that the modular unit (5, 105) has a power less than or equal to 200 KWatt, preferably less than or equal to 100 KWatt, even more preferably less than or equal to 20 KWatt.

24. Plant according to any one of claims 21 to 23, characterised in that at least one electrode (30, 130) is three-dimensional and has a section of greater size with respect to the other sections and intended to be arranged perpendicular to the direction of the flow (C) of water through the cell (7, 107*bis*, 107*ter*).

25. Plant according to any one of claims 21 to 24, characterised in that it lacks parts for total holding of the water around the electrodes (30, 130), for example it can comprise one or more channels to define a predetermined flow of water from and to the electrodes and/or one or more selective membranes (116).

26. Plant according to any one of claims 21 to 25, characterised in that it comprises at least one device (170) for detect-

ing at least one parameter of the water and/or at least one operating parameter of the at least one cell (7, 107 bis, 107 ter), said detection device (170) being in communication with the station (150) so as to form a feedback system.

27. Plant according to claim 26, characterised in that the feedback is of the adaptive type.

28. Plant according to any one of claims 21 to 27, characterised in that it comprises a plurality of devices generating at least one electrocapturing phenomenon (5, 105), and a control station (280), remote with respect to the place of application of the electrolysis (10), connected to said electrolytic generator devices through a telecommunication system (155).

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